

Open Access Article

Al/Al₂O₃ Nanoparticles Prepared by Pulsed Laser Ablation in Liquid

Abbas M. Ali Al-Kifaie, Ali Abid Abojassim*

Department of Physics, Faculty of Science, Kufa University, Iraq

Abstract: This research aims to study the optical properties of Al₂O₃ nanoparticles using pulsed laser beams. Pulsed laser ablation in liquid is cheap. We considered the optical properties of Al/Al₂O₃ nanoparticles obtained by pulsed laser ablation in liquid. An Nd:YAG RG190 laser with a wavelength of 1064 nm and focus energy of 950 mJ was used. The recurrence was 5 Hz, and the number of pulses was 300. The experiment was conducted at 25°C. A high-purity metallic plate was submerged in distilled water for the experiment. We found that the transmittance rises to 95%, which leads to a reduction in the absorbance and energy band gap. It can be concluded that pulsed laser ablation in liquid produces Al/Al₂O₃ nanoparticles with good optical properties.

Keywords: pulsed laser ablation, colloidal nanoparticles, surface plasmon resonance, Fourier transform infrared spectroscopy, field-emission scanning electron microscopy.

液体中脉冲激光烧蚀制备铝/氧化铝纳米颗粒

摘要: 本研究旨在使用脉冲激光束研究氧化铝纳米颗粒的光学特性。液体中的脉冲激光烧蚀很便宜。我们考虑了在液体中通过脉冲激光烧蚀获得的铝/氧化铝纳米颗粒的光学特性。使用波长为1064纳米且聚焦能量为950兆焦耳的钕:YAG RG190激光器。重复频率为5赫兹,脉冲数为300。实验在25°C下进行。将高纯金属板浸入蒸馏水中进行实验。我们发现透射率上升到95%,这导致吸光度和能带隙减小。可以得出结论,液体中的脉冲激光烧蚀产生具有良好光学性能的铝/氧化铝纳米颗粒。

关键词: 脉冲激光烧蚀、胶体纳米粒子、表面等离子体共振、傅里叶变换红外光谱、场发射扫描电子显微镜。

1. Introduction

Alumina (Al₂O₃) is commonly used to obtain metal oxide nanoparticles (NPs) for a few centuries now [1]. NPs can be modified by using coating techniques. Various coating materials have already been researched, but the preeminent common [2] for Al₂O₃. Al₂O₃ NPs are manufactured by solid-liquid blended casting combined with ultrasonic treatment [3]. The ultrasonic vibrations during the solidification were beneficial to refine the grain structure and to upgrade the final structure and properties of the Al₂O₃ NPs [4, 5]. Aluminum is one of first inexhaustible components discovered in soil and is widely utilized in numerous products due to its unique properties.

A high-purity metallic plate (98.88%) had a smooth surface and was first washed multiple times in an ultrasonic bath containing distilled water; then, it was rinsed with water 15 min before the start of the experiment. Colloidal NPs were obtained by using pulsed laser ablation in liquid (PLAL) using an Nd:YAG RG190 laser. The focus energy was 950 mJ, laser wavelength was 1064 nm, and number of pulses applied was 300, as shown in Fig. 1. The high-purity metallic (Al) plate was placed on the bottom of an open quartz vessel containing 3 ml of ultrapure distilled water; the space between the plate and laser was 12 cm.

2. Experimental Details

Received: May 1, 2021 / Revised: May 6, 2021 / Accepted: July 19, 2021 / Published: September 30, 2021

About the authors: Abbas M. Ali Al-Kifaie, Ali Abid Abojassim, Department of Physics, Faculty of Science, Kufa University, Iraq

Corresponding author Ali Abid Abojassim, ali.alhameedawi@uokufa.edu.iq

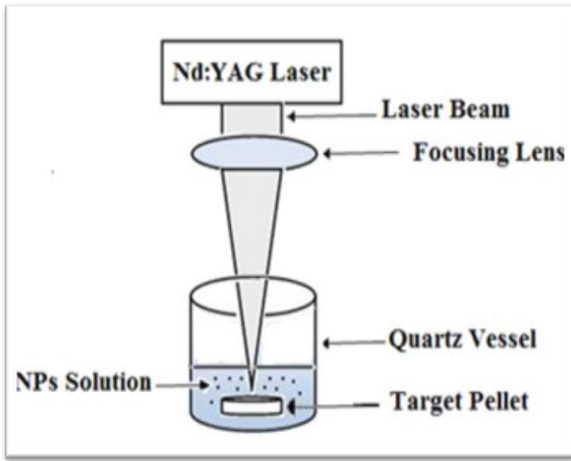


Fig. 1 Laser ablation system

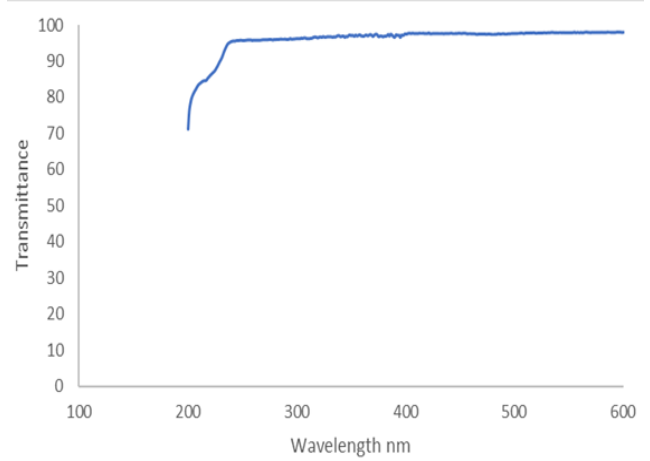


Fig. 3 Transmittance optical spectra of Al₂O₃Nps at 950 mJ

3. Results and Discussion

3.1. Optical Absorbance

An overview of the fundamental absorption bands of the conductor materials based on their absorbance spectra, transmittance details, and estimation can provide us with the optical properties, such as the energy band gap and UV–Vis absorbance spectra at room temperature (25°C).

3.2. Absorbance

The natural absorption band moved toward the shorter wavelength region. Surface plasmon resonance (SPR) was observed at 950 mJ and 217 nm, as shown in Fig. 2.

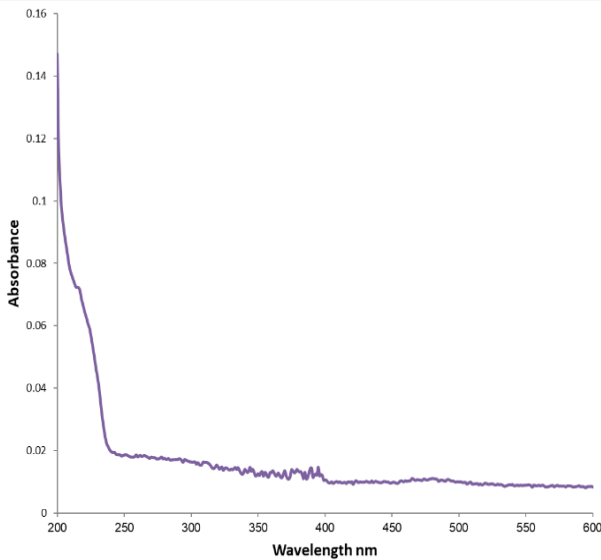


Fig. 2 Absorbance optical spectra of Al₂O₃Nps at 950 mJ

3.3. Transmittance

The transmittance band moved toward the longer wavelength region. SPR was observed at 950 mJ and 240 nm, as shown in Fig. 3.

3.4. Energy Band Gap

As shown in Fig. 4, the optical energy band gap for permitted coordinate move of the arranged colloidal NPs is evaluated using the Tauc plot, i.e., $(\alpha h\nu)^{1/2}$ versus $h\nu$. The optical band gap is significantly impacted by the chemical bonds of the components within the structure. Moreover, the band gap is influenced by the particle size.

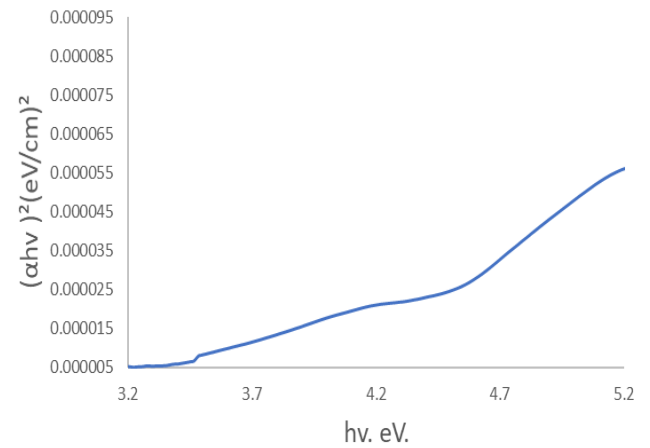


Fig. 4 The optical energy gap of Al₂O₃Nps at 950 mJ

Table 1 Optical properties of Al₂O₃ Nps

Metal	Energy(mJ)	Wavelength(nm)	Absorbance %	transmittance%	Energy gap (eV)
Al ₂ O ₃	950	1064	2	95	4.2

3.5. Field-Emission Scanning Electron Microscopy (FESEM)

The surface morphology of the obtained NPs is essential to analyze the shape of the obtained NPs and to study the NPs obtained in further detail. FESEM is employed for this. Fig. 5 shows the FESEM micrographs of the NPs, highlighting that the entire surface of the Si substrate is uniform. The NPs are completely developed, and their diameters are around 38.15–83.7 nm.

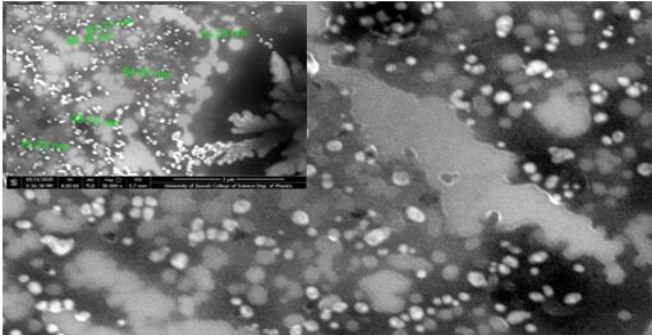


Fig. 5 FESEM of Al₂O₃ Nps at 950 mJ

3.6. Fourier Transform Infrared (FTIR) Spectroscopy

At wavelengths of 1450–4000 cm⁻¹, the FTIR spectra show the absorption results of only two atoms and not the absorption results of the entire molecule, as shown in Fig. 6. As the energy increases, the wavenumber also increases. Therefore, the bond energy increases with the transmittance, whereas the thickness decreases.

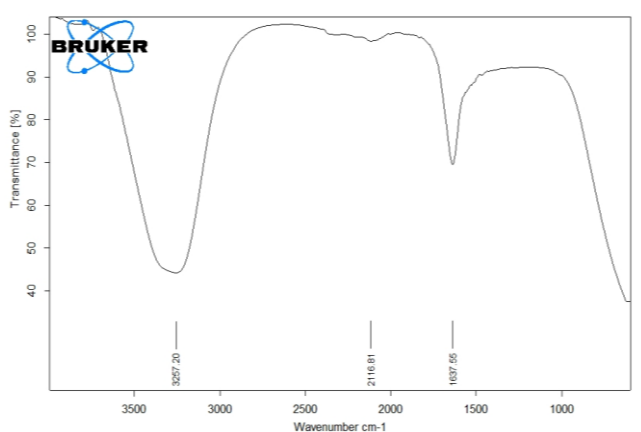


Fig. 6 FTIR of Al₂O₃ nanoparticle by pulsed laser ablation in 950 mJ

4. Conclusion

The laser removal strategy is quick, clean, and cheap, and has an amplified period of solidness. This could be a superior strategy to fabricate the internal structure of metal oxide and metal NPs. Manufacturing Al₂O₃ NPs by solid–liquid blended casting combined with a pulsed laser was beneficial to refine the grain structure and to upgrade the final structure and properties of the Al₂O₃ NPs. Finally, PLAL allows an amplified period of solidness and is cheap for conducting studies on Al/Al₂O₃ NPs, which have taken place for the first time in Iraq.

References

- [1] SALEH M., AL-HAJRI Z., POPELKA A., and JAVAID ZAIDI S. Preparation and characterization of alumina HDPE composites. *Materials*, 2020, 13(1): 250. <https://doi.org/10.3390/ma13010250>
- [2] AKIRA M. W., HARITSAH H., ZULFIA A., and PRAJATELISTIA E. Mechanical and Tribological Properties of Nano-Sized Al₂O₃ Particles on ADC12 Alloy Composites with Strontium Modifier Produced by Stir

Casting Method. *ADI Journal on Recent Innovation*, 2021, 3(1): 9-20. <https://doi.org/10.34306/ajri.v3i1.346>

[3] BARAKAT W., ELKADY O., ABUOQAIL A., YEHYA H., and EL-NIKHAILY A. Effect of Al₂O₃ coated Cu nanoparticles on properties of Al/Al₂O₃ composites. *Journal of Petroleum and Mining Engineering*, 2020, 22(1): 1-8. <https://doi.org/10.21608/JPME.2020.19110.1017>

[4] HOSSEINZADEH M., MIRZAEI O., and MOHAMMADIAN-SEMNANI H. Evaluation of microstructural and mechanical properties of A356 composite strengthened by nanocrystalline V8C7-Al₂O₃ particles synthesized through mechanically activated sintering. *Journal of Alloys and Compounds*, 2019, 782: 995-1007. <https://doi.org/10.1016/j.jallcom.2018.12.150>

[5] KRAUSE B. C., KRIEGEL F. L., ROSENKRANZ D., DREIACK N., TENTSCHERT J., JUNGNICHEL H., JALILI P., FESSARD V., LAUX P., and LUCH A. Aluminum and aluminum oxide nanomaterials uptake after oral exposure - a comparative study. *Scientific Reports*, 2020, 10(1): 2698. <https://doi.org/10.1038/s41598-020-59710-z>

参考文献:

[1] SALEH M., AL-HAJRI Z., POPELKA A. 和 JAVAID ZAIDI S.

氧化铝高密度聚乙烯复合材料的制备和表征。材料, 2020, 13(1): 250. <https://doi.org/10.3390/ma13010250>

[2] AKIRA M. W., HARITSAH H., ZULFIA A. 和 PRAJATELISTIA E.

纳米尺寸氧化铝颗粒在模数转换器12合金复合材料上的机械和摩擦学特性, 其中含有通过搅拌铸造法生产的镓改性剂。ADI 近期创新期刊, 2021, 3(1): 9-20.

<https://doi.org/10.34306/ajri.v3i1.346>

[3] BARAKAT W., ELKADY O., ABUOQAIL A., YEHYA H. 和 EL-NIKHAILY A. 氧化铝涂覆的铜纳米颗粒对铝/氧化铝复合材料性能的影响。石油与采矿工程学报, 2020, 22(1): 1-8. <https://doi.org/10.21608/JPME.2020.19110.1017>

[4] HOSSEINZADEH M., MIRZAEI O. 和 MOHAMMADIAN-SEMNANI H. 评估通过机械活化烧结合成的纳米晶V8C7-氧化铝颗粒强化的一种356

复合材料的微观结构和机械性能。合金与化合物杂志, 2019, 782: 995-1007. <https://doi.org/10.1016/j.jallcom.2018.12.150>

[5] KRAUSE B. C., KRIEGEL F. L., ROSENKRANZ D., DREIACK N., TENTSCHERT J., JUNGNICHEL H., JALILI P., FESSARD V., LAUX P. 和 LUCH A. 口服暴露后铝和氧化铝纳米材料的吸收-比较研究。科学报告, 2020, 10(1): 2698. <https://doi.org/10.1038/s41598-020-59710-z>